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DETERMINATION OF BISMUTH IN COPPER CONCENTRATE BY ATOMIC ABSORPTION SPECTROMETRY

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ABSTRACT

A method for bismuth determination in copper concentrate by atomic absorption spectrometry is presented. Investigations of the matrix interferences showed that iron increasing the bismuth absorbance. For this reason, a mathematical relation for correction of the obtained value for bismuth in dependence of iron content, is proposed.

INTRODUCTION

It is very important to follow the presence of bismuth in copper concentrate because in the copper production, bismuth passes into a metal phase, and it contaminates the obtained copper. The bismuth determination in geological materials is performed by different methods, but atomic absorption spectrometry (AAS) is one of the most applied one. In addition, different technique of atomic absorption spectrometry (AAS) are used: flame (ENDO et al., 1969; TARASEVICH et al., 1975; CLARK and VIETS, 1981; GUSKINSKII et al., 1982), electrothermal (LANGMYHR et al., 1974; KANE, 1979; HEINRICH, 1979; SCHROEN et al., 1983), and lately, hydride generation technique (DEKERSABIEC, 1980; TERASHIMA, 1984; CROCK, 1989; ZHANG, 1990).

In this work a method for direct bismuth determination in copper concentrate by flame AAS is presented.

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EXPERIMENTAL

Instrumentation

A Perkin-Elmer M 5000 atomic absorption spectrophotometer was applied. A bismuth hollow cathode lamp was used as a source, while deuterium lamp was used as a background corrector. The instrumental parameters are given in Table 1.

TABLE 1. Instrumental parameters for bismuth determination by AAS

Wavelength	223.1
Slit	0.2 nm
Lamp current	10 mA
Gas mixture	Air-Acetylene

Procedure

5 g of copper concentrate were taken into a beaker of 400 cm³ and 45 cm³ concentrated HCl and 15 cm³ concentrated HNO₃ were added. This solution was heated 30-40 minutes on a hot plate. Then, 25 cm³ distilled water were added and the solution was heated for another 15 minutes. The solution was cooled and filtered off. The filtrate was collected in a volumetric flask of 100 cm³, and filled with distilled water; from this solution the determination of bismuth by AAS was performed.

RESULTS AND DISCUSSION

In the application of AAS it is necessary to check the possible interferences on the determination of the investigated element, especially when the sample is of a complex composition. To avoid these interferences in the bismuth determination by flame AAS in different geological samples a matrix modification (Endo et al., 1969) or extraction of bismuth (TARASEVICH et al., 1975; CLARK and VIETS, 1981; GUSKINSKII et al., 1982; LANGMYHR et al., 1974) were applied. It was necessary to see the eventual influence of the elements with higher concentration in copper concentrates (copper, iron, calcium and lead) and in the obtained sample solution, on the bismuth determination. For this purpose a series of solutions with constant concentration of bismuth (10 mg. cm⁻³) and different concentration of the potential interfering elements, were prepared. The investigations show that the presence of calcium and lead (from 0 to 250 mg. cm⁻³, corresponding to 0.1 to

0.5 % in the solid sample) and copper (up to 16 mg. cm⁻³, corresponding to the concentration of 32 %) don't interfere on the absorption of bismuth.

However, the presence of iron in the solution (from 7-16 mg. cm⁻³, corresponding to the iron concentration of 24-32 %) increase the absorption of bismuth. It was found that there is a linear dependency on the increasing of the obtained bismuth concentration from the iron concentration, Fig. 1. Applying the linear regression analysis, the relation for the calculation of the correct value of the bismuth concentration was obtained from the found concentration of bismuth and the concentration of iron:

$$C_{\text{Bi}} = C_{\text{Bi}}(\text{found}) - 0.06212 \cdot C_{\text{Fe}} - 0.3126$$

were the corresponding concentrations are given in $\mu\text{g. cm}^{-3}$ for C_{Bi} , and in mg. cm⁻³ for C_{Fe} .

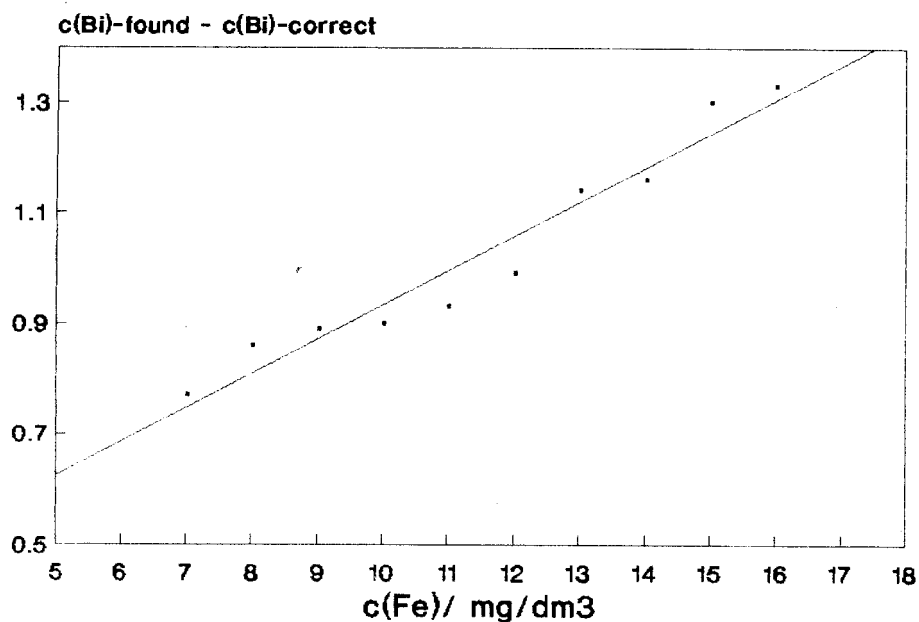


Fig. 1. The dependence of difference between C_{Bi} (found) and C_{Bi} (correct) from the iron concentration

To cheque this relation, three samples of copper concentrate were analyzed and the correct values of bismuth were calculated, and than these values were

compared with the values obtained by the spectrographic method, as shown in Table 2. One can notice that the corrected values of bismuth are similar to those obtained by spectrographic method.

TABLE 2. Results from the bismuth determination in copper concentrate samples by proposed method and by spectrographic method

Sample No	w_{Fe} (%)	Proposed method (%) w_{Bi} (found)	Spectrographic method (%) w_{Bi} (corrected)
L-1	31.81	0.089	0.079
L-3	29.75	0.165	0.147
L-11	30.05	0.160	0.142

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РЕЗИМЕ

ОПРЕДЕЛУВАЊЕ НА БИЗМУТ ВО БАКАРЕН КОНЦЕНТРАТ СО ПОМОШ НА АТОМСКА АПСОРПЦИОНА СПЕКТРОМЕТРИЈА

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Во трудот е прикажана постапка за определување на бизумот во бакарни концентрати со помош на атомска апсорпциона спектрометрија. Испитувањата на влијанието на присутниот матрикс покажа дека железото влијае врз определувањето на бизумот со тоа што ја зголемува вредноста на апсорбанцата. Предложена е математичка корекција на добиената вредност на концентрацијата на бизмут во зависност од концентрацијата на железо во добиениот раствор.

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REFERENCES

- GUSKINSKII, M. N., PASHADZHANOV, A. N. NEMODRUK, A. A. 1982. Use of water-miscible organic solvents in the extraction-atomic absorption determination of busmuth in rocks, ores and leadzinc concentrates. *Zh. Anal. Khim.*, 31, 42-48.
- DEKERSABIEC, A. M., 1980. Determination of arsenic, selen and bismuth in rocks and soils by atomic absorption spectrometry. *Analisis*, 8, 97-101.
- ZHANG, P., 1989. Determination of arsenic, antimony and bismuth in geological samples by hydride generation AAC. *Guangpuxue, Yu Guangpu Fenxi Zazhi*, 9, 329.
- ENDO, Y., HARA, T., NAKAHARA, V. 1969. Determination of Ti, Ni, Cr, Pb and Bi in iron ore by atomic absorption spectrometry. *Bunseki Kagaku*, 18, 833-8.
- KANE, J. S. 1979. Determination of nanogram amounts of bismuth in rocks by atomic absorption spectrometry with electrothermal atomization. *Anal. Chim. Acta*, 106, 325-31.
- LANGMYHR, F. J. SOLBERG, R., WOLD, L. T., 1974. Atomica absorption spectrometry determination of silver, bismuth and cadmium in sulfide ores by direct atomization from the solid state. *Anal. Chim. Acta*, 69, 267-73.
- SCHROEN, W., BOMBACH, G., BEUGE, P., 1983. Rapid flameless atomic absorption spectrometry study of trace elements in geological samples. *Spectrochim. Acta*, 38B (10), 1269-76.
- TARASEVICH, N. I., KOZYREVA, G. V., PORTUGALSKAYA, Z. P., 1975. Extraction-atomic absorption determination of indium, bismuth and lead trace impurities in rocks and soils. *Vestn. Mosk. Univ., Khim.*, 16(2), 241-3.
- HEINRICH, H., 1979. Determination of bismuth, cadmium and thallium in 33 international standard reference rocks by fractional distillation combined with flameless atomic absorption spectrometry. *Fresenius Z. Anal. Chem.*, 294, 345-51.
- TERASHIMA, S., 1984. Determination of bismuth in geological materials by automated hydride generation and electrothermal atomic absorption spectrometry. *Anal. Chim. Acta*, 156, 301-5.
- CROCK, J. G., 1989 An improved method for the determination of bismuth in geological materials by automated hydride generation AAS. *Anal. Lett.*, 19, 1367-85.
- CLARK, J. R., VIETS, J. G. 1981. Multielement extraction system for the determination of 18 trace elements in geological samples. *Anal. Chem.*, 53, 61-5.